South Plains College Common Course Syllabus: PHYS 1410 Revised 06/30/2023

Department: Science Discipline: Physics Course Number: PHYS 1410 Course Title: Elementary Physics Available Formats: conventional Campuses: Levelland Instructor: David Hobbs Office: S67 Office Hours: MW 8:30 – 11:00 am, F 8:30 – 11:30 am Phone: 806-716-2639 email: dhobbs@southplainscollege.edu

Course Description: Conceptual level survey of topics in physics intended for liberal arts and other non-science majors.

Prerequisite: There are no prerequisites for this course, however you will be expected both on the homework and in the exams to be able to perform simple mathematical calculations. Examples of the mathematical concepts we will use in this course are scientific notation, multiplying and dividing powers of 10, converting between different metric units, rearranging and solving simple equations. It will be assumed that you are familiar with high school algebra.

Credit: 4 Lecture: 3 Lab: 3

Textbook: Conceptual Physics, 13th Edition by Paul G. Hewitt (Pearson, 2021). The textbook and Mastering Physics learning platform will be available through Blackboard.

Supplies: Scientific Calculator

This course partially satisfies a Core Curriculum Requirement:

Life and Physical Sciences Foundational Component Area (030)

Core Curriculum Objectives addressed:

- **Communications skills**—to include effective written, oral and visual communication
- **Critical thinking skills**—to include creative thinking, innovation, inquiry, and analysis, evaluation and synthesis of information
- **Empirical and quantitative competency skills**—to manipulate and analyze numerical data or observable facts resulting in informed conclusions
- **Teamwork**—to include the ability to consider different points of view and to work effectively with others to support a shared purpose or goal

Student Learning Outcomes:

Learning Outcomes - Upon successful completion of this course, students will:

- 1. Distinguish between displacement, velocity, and acceleration
- Solve simple problems involving uniform motion, uniformly accelerated motion, or uniform circular motion
- 3. State Newton's Laws of Motion, explain the meaning of each, and identify applications of each
- 4. Apply Newton's laws of motion to relate forces to motion for simple physical cases
- 5. Define momentum and solve simple problems involving conservation of momentum
- Identify types of energy in a system and solve simple problems involving conservation of energy
- 7. Describe the basic structure of an atom in terms of protons, neutrons, and electrons
- 8. Make simple calculations involving changes in temperature as well as phase changes when systems at different temperatures interact
- 9. Describe and calculate basic properties of waves such as frequency, wavelength, and amplitude
- 10. Discuss wave interference and the conditions for constructive and destructive interference
- 11. Describe standing waves and determine the frequencies of the harmonics
- 12. Discuss electric charge and the role it plays in atomic structure.
- 13. Calculate electrical forces using Coulomb's law.
- 14. Describe electric field and discuss electrical interactions in terms of electric field.
- 15. Discuss simple electrical circuits and make calculations using Ohm's law applied to series and parallel circuits.
- 16. Describe magnetic field and discuss interactions of magnetic fields with moving charges.
- 17. Discuss and apply Faraday's law of electromagnetic induction
- 18. Relate changing magnetic fields to induced electric fields.
- 19. Describe electromagnetic waves in terms of electric and magnetic fields and electromagnetic induction
- 20. Discuss the spectrum of electromagnetic waves from radio waves to x-rays.
- 21. Discuss diffraction and interference and how they arise based on superposition and Huygens' Principle.
- 22. Make simple calculations related to the photoelectric effect and the Bohr model of the hydrogen atom
- 23. State the Pauli Exclusion Principle and specify its implications for atomic structure
- 24. Discuss how quantum mechanics explains the structure of the periodic table
- 25. Describe the basic structure of a nucleus and explain the meaning of different isotopes
- 26. Recall the three basic types of radioactivity and describe some properties of each
- 27. Use radioactive half-life in simple calculations
- 28. Describe the basic principles of radioactive dating
- 29. Discuss the use of nuclear fission in electric power generation

Student Learning Outcomes Assessment: Selected questions on tests will assess how well students have met targeted student learning outcomes.

Course Evaluation: Student grades will be based on daily work, homework, and tests. Final grades will be assigned based on the percentages shown below:

Task	Weight
Daily Work	25%
HW & Tests	75%

The letter grades will be based on a fixed scale as follows:

A: 89.5 - 100 B: 79.5 - 89.5 C: 69.5 - 79.5 D: 59.5 - 69.5 F: below 59.5 Borderline cases (within 0.5 of the break) will be decided based on class participation.

Attendance Policy: Attendance and effort are vital to success in this course. Class attendance keeps you well connected to the course and gives you opportunities to ask questions and clear up confusions. Therefore, students are expected to be in attendance for every class session. Students with excessive absences (more than 5) will be administratively dropped from the class. It is the student's responsibility to know how many absences they have accumulated.

Daily Work: Daily work consists of reading quizzes and in-class (lab) practice with feedback. These activities are meant to be formative exercises and are graded primarily on participation. Their purpose is to help develop understanding of the concepts and principles and to prepare you for the tests.

Daily Work Grade Determination: 50% of your daily work grade will come from the reading quizzes and 50% from the in-class practice.

Homework: Do your homework! There is no substitute. Students who don't put in a good effort often struggle in the course. Homework will be assigned and graded online. A better semester average homework grade will replace your lowest test score.

Tests: Three tests will be given during the semester as shown on the course calendar. Each test will be worth 25% of the course grade. There will be no make-up tests given, so a test missed counts as zero. However, your lowest test grade will be <u>replaced automatically</u> by a greater semester average homework score at the end of the semester. Thus, in addition to demonstrating your grasp of the subject and helping you to prepare for tests, a good homework grade provides "insurance" against a low or missing test grade.

Plagiarism and Cheating: Students are expected to do their own work on all projects, quizzes, assignments, examinations, and papers. Failure to comply with this policy will result in an F (grade of zero) for the assignment and can result in an F for the course if circumstances warrant.

Plagiarism violations include, but are not limited to, the following:

- 1. Turning in a paper that has been purchased, borrowed, or downloaded from another student, an online term paper site, or a mail order term paper mill;
- 2. Cutting and pasting together information from books, articles, other papers, or online sites without providing proper documentation;
- 3. Using direct quotations (three or more words) from a source without showing them to be direct quotations and citing them; or
- 4. Missing in-text citations.

Cheating violations include, but are not limited to, the following:

- 1. Obtaining an examination by stealing or collusion;
- 2. Discovering the content of an examination before it is given;
- 3. Using an unauthorized source of information (notes, textbook, text messaging, internet, apps) during an examination, quiz, or homework assignment;
- 4. Entering an office or building to obtain unfair advantage;
- 5. Taking an examination for another;
- 6. Altering grade records;
- 7. Copying another's work during an examination or on a homework assignment;
- 8. Rewriting another student's work in Peer Editing so that the writing is no longer the original student's;
- 9. Taking pictures of a test, test answers, or someone else's paper.

Student Code of Conduct Policy: Any successful learning experience requires mutual respect on the part of the student and the instructor. Neither instructor nor student should be subject to others' behavior that is rude, disruptive, intimidating, aggressive, or demeaning. Student conduct that disrupts the learning process or is deemed disrespectful or threatening shall not be tolerated and may lead to disciplinary action and/or removal from class.

For information regarding official South Plains College statements about intellectual exchange, disabilities, non-discrimination, Title IX Pregnancy Accommodations, CARE Team, and Campus Concealed Carry, please visit

https://www.southplainscollege.edu/syllabusstatements/.

Note: The instructor reserves the right to modify the course syllabus and policies, as well as notify students of any changes, at any point during the semester.

Calendar

., <u>,</u> , , ,	Tuesday		Thursday	
Week	Readings	Topics	Readings	Topics
1	08/29	Course Intro – Blackboard, Mastering Physics; Review of Scientific Notation, SI Units	08/31	What is Science? The Scientific Attitude
			Ch1	
2	09/05	Motion is Relative – "At Rest" and "Moving Uniformly" are Dynamically the Same	09/07	Acceleration is Changing Motion
	Ch2		Ch3	
3	09/12	Forces Cause Motion to Change	09/14	Forces Mediate Interactions
	Ch4		Ch5	
4	09/19	Momentum is Conserved	09/21	Energy is Conserved
	Ch6		Ch7	
5	09/26	Heat is Energy Transfer That Changes Thermal Energy	09/28	Angular Momentum is Conserved
	Ch15/17		Ch8	
	10/03	Gravity is Universal	10/05	Test 1 Chapters 1 – 8, 15, 17
6	Ch9			
7	10/10	Matter is Made of Atoms	10/12	Properties of Solids, Liquids, and Gases
	Ch11		Ch12-14	
8	10/17	Vibrations and Waves	10/19	Electric Force is Mediated by Electric Field
	Ch19		Ch22	
9	10/24	Practical Electricity – Electric Circuits	10/26	Magnetic Force is Mediated by Magnetic Field
	Ch23		Ch24	
10	10/31	Light is an Electromagnetic Wave	11/02	Interaction of Light with Matter – Reflection and Refraction
	Ch25 Ch26		Ch28	
11	11/07	Light is Wave-like – Diffraction and Interference	11/09	Test 2 Chapters 9, 11-14, 19, 22 – 26, 28
11	Ch29			
12	11/14	Interaction of Light with Matter – Incandescence; Emission and Absorption Spectra	11/16	Light is Particle-like – Photoelectric Effect Particles are Wave-like – de Broglie Waves
	Ch30		Ch31	
13	11/21	Quantum Mechanics Explains Atomic Structure	11/23	Thanksgiving – No Class
	Ch32			
14	11/28	Strong and Weak Nuclear Interactions	11/30	Energy from the Nucleus
	Ch33		Ch34	
15	12/05	Space and Time => Spacetime	12/07	Spacetime is Curved and Dynamic
	Ch35		Ch36	
16	12/12	Test 3 Chapters 29 – 36 1:00 – 3:00 pm	12/14	